Acta Cryst. (1998). C54, 1145-1146

3-tert-Butyl-2'-chloro-2-hydroxy-5-methyl-azobenzene

Şamil İşık, a Muhittin Aygün, b Hasan Kocaokutgen c and M. Nawaz Tahır d

^aDepartment of Physics, Ondokuz Mayis University, TR-55139, Samsun, Turkey, ^bDokuz Eylul University, Faculty of Buca Education, Department of Physics, 35510-Buca, Izmir, Turkey, ^cDepartment of Chemistry, Ondokuz Mayis University, TR-55139, Samsun, Turkey, and ^dDepartment of Physics Engineering, Hacettepe University, Beytepe 06532, Ankara, Turkey. E-mail: samili@samsun.omu.edu.tr

(Received 11 November 1997; accepted 17 February 1998)

Abstract

The crystal structure of the title compound, 6-tert-butyl-2-(2-chlorophenyldiazenyl)-4-methylphenol, $C_{17}H_{19}Cl-N_2O$, determined at room temperature, shows that the molecule is essentially planar, as found for other azobenzene derivatives, and features a significant intramolecular $H\cdots N$ interaction of 1.75 (3) Å resulting in the formation of a six-membered ring.

Comment

The synthesis and spectroscopic characterization of the title compound, (I), has been reported previously (Kocaokutgen & Gumrukcuoglu, 1996). The present report details the X-ray crystal structure of (I), which was undertaken in order to determine the conformation of the molecule.

$$CI$$
 HO CH_3 CH_3 CH_3 CH_3 CH_3

A view of a molecule of (I), with the atom labelling, is shown in Fig. 1. Bond lengths and angles associated with the phenyl portions of the molecule are as expected. The N2—C1 and N1—C7 bond lengths of 1.405 (3) Å and 1.417 (3) Å, respectively, indicate single-bond character, a result consistent with those found for other azobenzene derivatives studied recently (Işik *et al.*, 1997, 1998). The N=N bond length of 1.266 (2) Å is indicative of significant double-bond character; the C—Cl bond distance is 1.735 (3) Å.

The aromatic rings, which adopt a *trans* configuration about the N=N double bond, are essentially planar. The dihedral angle between the C1—N2=N1—C7 azo moiety and the chloro-substituted ring is 1.88 (4)°, that between the azo group and the other aromatic ring is

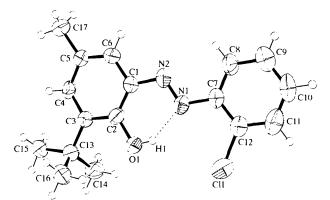


Fig. 1. ORTEPII (Johnson, 1976) view of the title molecule showing the atomic numbering scheme. The displacement ellipsoids are at the 50% probability level and H atoms are shown as spheres of arbitrary size.

 $3.98(1)^{\circ}$, and that between the two rings is $5.85(4)^{\circ}$. These results indicate that the molecule as a whole is essentially planar.

A significant intramolecular interaction involving the hydroxy H1 atom and the N1 atom is noted such that a six-membered ring is formed [H1···N1 1.75 (3), O1···N1 2.54 (3) Å and O1—H1···N1 148(2)°].

Experimental

The title compound was obtained as described previously (Kocaokutgen & Gumrukcuoglu, 1996) and recrystallized from ethanol to produce crystals of suitable quality for X-ray diffraction analysis.

Crystal data

- 3	
$C_{17}H_{19}CIN_2O$ $M_r = 302.79$ Triclinic $P\overline{1}$ $a = 7.1610(1) \text{ Å}$ $b = 9.445(2) \text{ Å}$ $c = 12.479(3) \text{ Å}$ $\alpha = 80.630(2)^{\circ}$ $\beta = 72.441(3)^{\circ}$ $\gamma = 83.272(2)^{\circ}$	Mo $K\alpha$ radiation $\lambda = 0.71069 \text{ Å}$ Cell parameters from 25 reflections $\theta = 11.31-20.83^{\circ}$ $\mu = 0.242 \text{ mm}^{-1}$ T = 293 (2) K Prismatic $0.35 \times 0.25 \times 0.20 \text{ mm}$ Red
	Red

Data collection

Enraf–Nonius CAD-4
diffractometer
ω/θ scans
Absorption correction: none
3418 measured reflections
3158 independent reflections
2411 reflections with
$I > 2\sigma(I)$

 $R_{\rm int} = 0.010$ $\theta_{\rm max} = 26.85^{\circ}$ $h = 0 \rightarrow 8$ $k = -11 \rightarrow 11$ $l = -14 \rightarrow 15$ 3 standard reflections frequency: 120 min intensity decay: 0.3%

Refinement

Refinement on F^2 $(\Delta/\sigma)_{\text{max}} = 0.001$ $\Delta \rho_{\text{max}} = 0.25 \text{ e Å}^{-3}$ $R[F^2 > 2\sigma(F^2)] = 0.055$ $wR(F^2) = 0.140$ $\Delta \rho_{\min} = -0.28 \text{ e Å}^{-3}$ Extinction correction: none S = 1.097Scattering factors from 3158 reflections International Tables for 266 parameters All H atoms refined Crystallography (Vol. C) $w = 1/[\sigma^2(F_o^2) + (0.0485P)^2]$ + 0.3914Pwhere $P = (F_o^2 + 2F_c^2)/3$

Table 1. Selected geometric parameters (Å, °)

CI1—C12	1.735(3)	C3—C13	1.537 (3)
O1—C2	1.345 (3)	C5C17	1.506 (3)
N2N1	1.266 (2)	C13—C15	1.532(3)
N2—C1	1.405 (3)	C13—C14	1.535 (4)
NI—C7	1.417 (3)	C13—C16	1.536 (4)
C2-O1-H1	106.1 (19)	C14C13C16	110.1 (2)
N1—N2—C1	115.8(2)	C15—C13—C3	111.8 (2)
N2—N1—C7	115.1(2)	C14—C13—C3	110.2 (2)
C15C13C14	106.7(2)	C16—C13—C3	109.7 (2)
C15C13C16	108.3 (2)		
C1-N2-N1C7	177.3 (2)	N2-N1-C7-C12	-179.0(2)
N1—N2—C1—C6	-179.4(2)	N2N1C7C8	-0.1(3)
N1—N2—C1—C2	-1.0(3)		

Data were corrected for Lorentz-polarization factors. The structure was solved by direct methods using *SHELXS*86 (Sheldrick, 1990) and refined by full-matrix least squares using *SHELXL*93 (Sheldrick, 1993). Non-H atoms were refined anisotropically. The H atoms were located from difference-Fourier maps and refined isotropically. C—H distances were in the range 0.88 (6)-1.03 (3) Å, and U_{iso} values for H atoms were in the range $0.044 (6)-0.18 (2) \text{ Å}^2$.

Data collection: *CAD-4 EXPRESS* (Enraf-Nonius, 1993). Cell refinement: *CAD-4 EXPRESS*. Data reduction: *MolEN* (Fair, 1990). Molecular graphics: *ORTEPII* (Johnson, 1976). Software used to prepare material for publication: *SHELXL93*.

The authors wish to acknowledge for the use of the CAD-4 diffractometer (purchased under Grant DPT/TBAG1 of the Scientific and Technical Research Council of Turkey) of the Physics Engineering Department, Hacettepe University, Turkey.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: FR1105). Services for accessing these data are described at the back of the journal.

References

Enraf-Nonius (1993). CAD-4 EXPRESS. Enraf-Nonius, Delft, The Netherlands.

Fair, C. K. (1990). MolEN. An Interactive Intelligent System for Crystal Structure Analysis. Enraf-Nonius, Delft, The Netherlands.
Işik, Ş., Aygün, M., Kocaokutgen, H., Tahir, M. N., Büyükgüngör, O. & Erdönmez, A. (1998). Acta Cryst. C54, 859-860.

Işik, Ş., Aygün, M., Şaşmaz, S., Kendi, E., Büyükgüngör, O. & Erdönmez, A. (1997). Acta Cryst. C53, 593-594.

Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA. Kocaokutgen, H. E. & Gumrukcuoglu, I. (1996). Spectrosc. Lett. 29, 185-192.

Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.

Sheldrick, G. M. (1993). SHELXL93. Program for the Refinement of Crystal Structures. University of Göttingen, Germany.

Acta Cryst. (1998). C54, 1146-1148

N-(2-Dimethoxymethyl-3-hydroxy-2-methyl-6-nitro-3,4-dihydro-2*H*-1-benzopyran-4-yl)-acetamide

Tae-Sung Yoon, a Sung Eun Yoo b and Whanchul Shin a

^aDepartment of Chemistry and Center for Molecular Catalysis, Seoul National University, Seoul 151-742, Korea, and ^bKorea Research Institute of Chemical Technology, PO Box 9, Daedeog-Danji, Daejeon 305-606, Korea. E-mail: nswcshin@plaza.snu.ac.kr

(Received 31 July 1997; accepted 6 January 1998)

Abstract

The benzopyran-based title compound, $C_{15}H_{20}N_2O_7$, is a potential potassium-channel opener with cardio-vascular therapeutic activities. The 2-dimethoxymethyl and 3-hydroxyl groups are in axial positions while the 4-acetamide group is in a pseudo-equatorial position. The 4-acetamide group is nearly perpendicular to the benzopyran plane. There are two intermolecular hydrogen bonds; one between the hydroxyl O and acetamide O atoms [2.725 (5) Å], and the other between the acetamide N and methoxy O atoms [3.139 (5) Å].

Comment

ATP-sensitive potassium-channel openers are promising in cardiovascular therapy as antihypertensives and coronary vasodilators and consist of several structural groups including benzopyran compounds such as cromakalim, (II) (Robertson & Steinberg, 1990). The title compound, (I), is one of a series of benzopyran-based